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Answer the

following questions

in the space

provided. 1.

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Following are statements about the dissolving process. Explain each one at the molecular level.

a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.

b.

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PROCESS Answers

to Review

Questions 13-2 The

inventory process

is affected by

control activities in

the revenue,

purchasing, and

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Solutions of Gases  
in Liquids Goals To  
describe the  
process by which  
gases dissolve in

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liquids. To explain why gases have a solubility limit in liquids. To explain why increased partial pressure of a gas over a liquid will lead to an increase in the solubility of that gas.

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and Colligative  
Properties MIXED  
Solutions REVIEW SHORT  
Section 2 ANSWER Answer

Answers  
the following  
questions in the  
space provided. 1.  
Match the four  
compounds on the  
right to their  
descriptions on the  
left. b an ionic  
compound that is  
quite soluble in

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water (a) HCl c an ionic compound that is not very soluble in water (b)  $\text{NaNO}_3$

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ANSWER Answer

the following

questions in the

space provided. 1.

Describe the errors

made by the



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Following students in making molar solutions. 3. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid. James made several errors.

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SHORT ANSWER 1.

c 2. a 3. b 2. a.

alcohol b. water c.

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the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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games  $\times$  310 g b  
am all e s = 34.  
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...

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