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For the levels to differ the pressure P_1 must be greater than P_2 , hence. $P_1 = P_2 + h \rho g$. If P_1 is the lung pressure, P_0 is the atmospheric pressure, then if the difference is ‘ h ’ then lung pressure can calculated as follows.. $P_1 = P_0 + h \rho g$. Example. A man blows into one end of a U-tube containing water until the levels differ by 40.0 cm. if the atmospheric pressure is 1.01×10^5 ...

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